

Adsorbents and Carbon Technology for Air Sampling & Thermal Desorption

PEFTEC November, 2015

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Agenda

- Introduction
- Type of Adsorbents
- Selection Tool



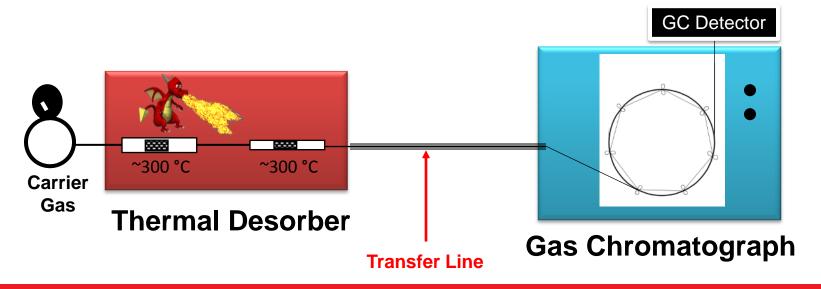




What is Thermal Desorption?

A Sample Preparation technique for Air & Gas Analysis and GC.

- The gas /air sample is collected onto an adsorbent packed glass or stainless steel tube. *The sample is concentrated on the adsorbents.*
- The packed tube is heated (Thermal) and the compounds are released into the carrier gas (Desorption) transfered onto GC column for Analysis





Adsorbent Characteristics for Thermal Desorption

- Able to retain & release the compounds of interest
- Able to withstand high temperatures ~ 300°C
- Low background levels
- Low metal content
- Hydrophobic

Desirable

- Consistent mesh/particle size
- Consistent density
- Low shrinkage
- Low amount of fines



Terms Defined

Surface Area:

- provides a general idea of the adsorbent strength, but it doesn't provide the whole picture.
- Other characteristics such as pore size, pore shape, and porosity can also play a role in the adsorbent's ability to retain and release different compounds.

<u>General Rule:</u>

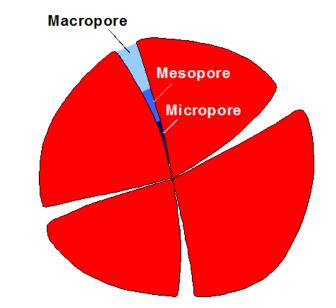
- The higher the surface area value, the stronger the adsorbent.
- For surface area >800 m²/g, the size and shape of the pore becomes more important.



Terms Defined (cont.)

Pore Size:

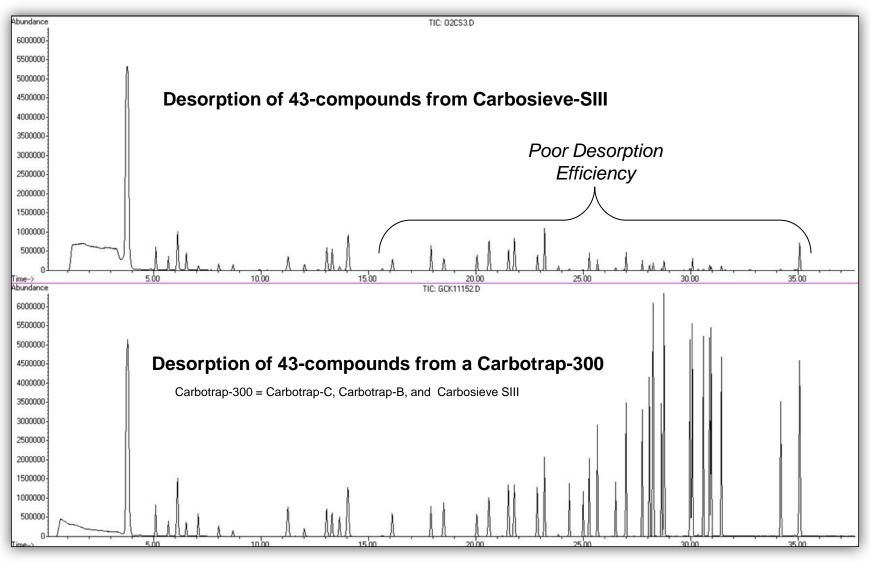
- Macropores: > 50 nm diameter
- Mesopores: 2 50 nm diameter
- Micropores: < 2 nm diameter



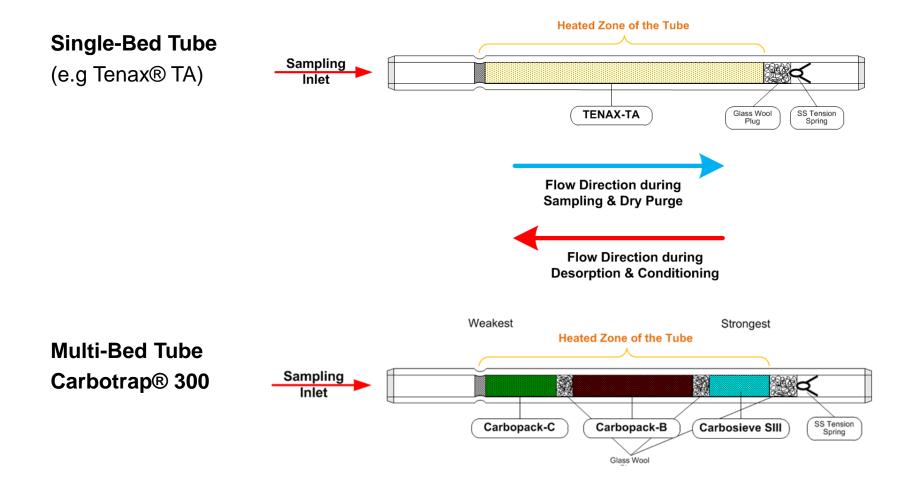
SIGMA-ALDRICH®

Molecules cannot access pores smaller than their size

Will the strongest adsorbent work for everything?



Thermal Desorption Tube Adsorbent Beds



Adsorbent Types used in Thermal Desorption



Typical Adsorbents for Thermal Desorption

- Polymers
 - Tenax[®] TA (2,6-diphenyl-p-phenylene oxide)
- Graphitized Carbon Blacks (GCB)
 - Non porous
 - Names: Carbopack[™], Carbotrap[™]
 - Various types available
- Carbon Molecular Sieves (CMS)
 - Porous
 - Names: Carboxen[™], Carbosieve[™]
 - Various types available
- Glass beads
 - Used to retain large molecular weight volatiles

Key Expertise of Sigma-Aldrich / Supelco !!

Materials used on the NASA missions Galileo (Jupiter) & Cassini-Huygens (Saturn-Titan)

Carbon Adsorbents History in Supelco

>20 year Experience in preparing Carbon Adsorbents

Carbosieves

- First Family of Carbon molecular sieves (S-I, S-II, S-III)
- Carbosieve S-III one of strongest adsorbens
 - Microporus only
 - First CMS in air monitoring

Carboxenes (Introduced 1987)

- Carboxene-1000 highly efficient adsorbent
 - Not as strong as S-III but far better kinetics
 - Materials on NASA Missions
 - (e.g. Galileo, Cassini/Huygens to Titan)

Adsorbent	BET	density		poros		pore
	surface area	(g/mL)	micro	- meso-	macro-	diameter
	(m2/g)		-po	ores (cc.	/g)	(Å)
Graphitized Car	bons					
Carbopack X	240	0.41	^	0.62	~	100
Carbopack Z	220	0.18	~	1.73	~	255
Carbopack B	100	0.35	~	~	~	N/A
Carbopack Y	24	0.42	^	^	^	N/A
Carbopack C	10	0.68	~	~	~	N/A
Carbopack F	5	0.64	^	^	^	N/A
Carbon Sieves						
Carboxen - 563	510	0.53	0.24	0.15	0.24	7 - 10
Carboxen – 564	400	0.60	0.24	0.13	0.14	6-9
Carboxen - 569	485	0.58	0.20	0.14	0.10	5 - 8
Carboxen - 1000	1200	0.48	0.44	0.16	0.25	10 - 12
Carboxen - 1001	500	0.61	0.22	0.13	0.11	5 - 8
Carboxen - 1002	1100	0.43	0.36	0.28	0.30	10 - 12
Carboxen - 1003	1000	0.46	0.38	0.26	0.28	5 - 8
Carboxen - 1006	715		0.29	0.26	0.23	7 - 10
Carboxen - 1010*	675	0.60	0.35	^	^	6 - 8
Carboxen - 1011	1100	0.48	0.41	0.19	0.24	10 - 12
Carboxen - 1012	1500	0.50	^	0.66	^	19 - 21
Carboxen - 1016	75	0.52	~	0.34	~	A
Carboxen - 1018	700	0.60	0.31	0.04	^	6 - 8
Carboxen - 1021	650	0.64	0.29	0.02	^	5 - 7
Carbosieve S-III	820	0.61	0.35	0.04	~	4 - 11
Carbosieve S-II	1059		0.45	0.01	^	6 - 15
Carbosieve G	1160		0.49	0.02	~	6-15
NASA 20/45	61	0.55	~	0.33	^	~
(Carboxen - 1017)					
Supelcarb	1150	0.46	0.47	0.26	0.28	5 - 8

Porous Polymers

Tenax®-TA

- The most popular adsorbent used in thermal desorption
- Maximum temperature: 350 °C
- Recommended desorption temp: 300 °C
- Recommended conditioning temp: 320 °C
- Methanol not retained (Good for spiking tubes with liquid calibration standards)

Typical Characteristics

- Granular Tan in color
- Surface area: 35 m²/g
- Hydrophobic





Porous Polymers (cont.)

PoraPak[™]-N, Chromosorb[®]-106, HayeSep[®]-D

- Relatively low maximum temperatures: 225-290 °C
- Recommended desorption temp: 200 °C
- Recommended conditioning temp: 210 °C
- Typically has higher background levels than other adsorbents

Typical Characteristics

- Spherical Light Yellow in color
- Surface area: 500 to 800 m²/g
- Hydrophobic

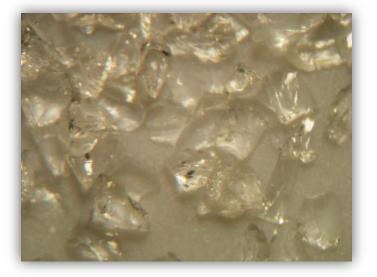


Other Adsorbents

Glass Beads 5 m²/g ("good pre-filter")

Rarely used for Thermal Desorption

- Petroleum Charcoal
- Coconut Charcoal
- Silica Gel
- Molecular Sieves (Zeolites)





Graphitized Carbon Blacks

Carbotrap[®] and Carbopack[™]

- Maximum temperature: 400 °C
- Recommended desorption temp: 330 °C
- Recommended conditioning temp: 350 °C
- Methanol not retained by most of them

Typical Characteristics

- Granular Flat Grey/Black in color
- Surface area: 5 to 240 m²/g
- Designed to retain and release mid to large molecular weight compounds
- Hydrophobic
- High Purity Low Background

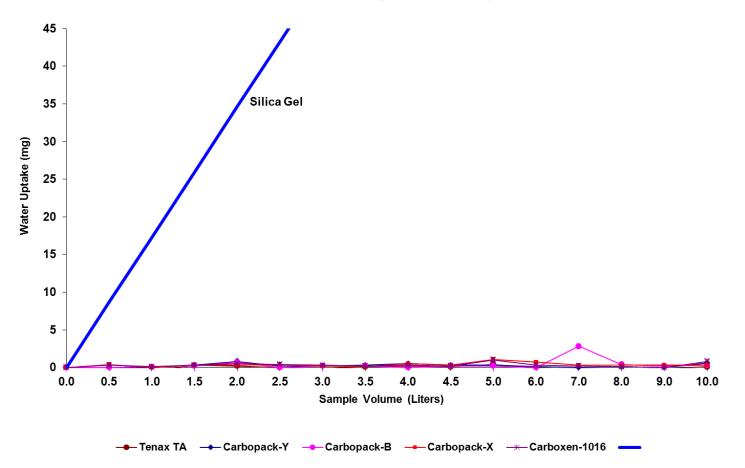
Carbotrap = 20/40 mesh **Carbopack** = 40/60, 60/80, 80/100, and 100/120 mesh



Water vapor retained by Carbopack(s) & Tenax-TA

Cumulative Water Uptake

Sampling at 75%RH (25°C)



Adsorptive Strength of Graphitized Carbon Blacks

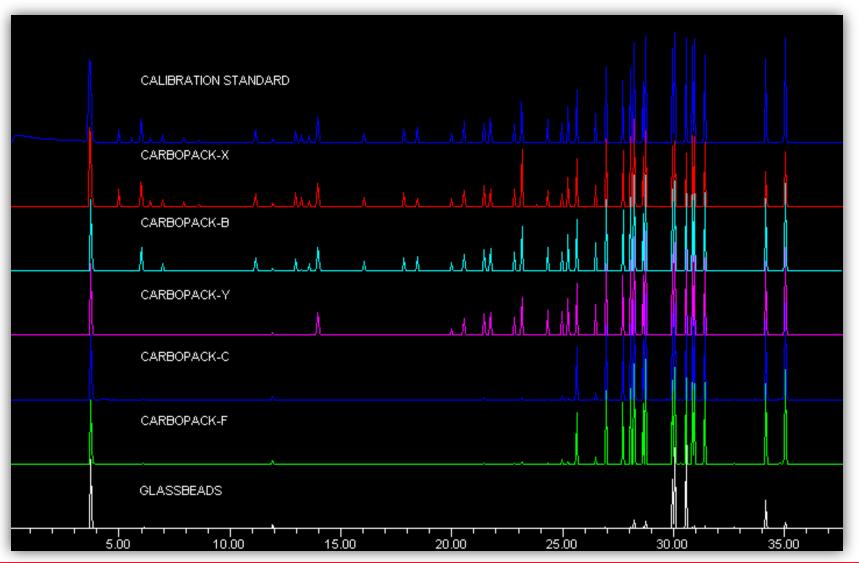


Carbopack X	240 m²/g	Strongest
Carbopack Z	220 m²/g	
Carbopack B	100 m²/g	
Carbopack Y	24 m²/g	
Carbopack C	10 m²/g	
Carbopack F	5 m²/g	Weakest

Relative Adsorption Strength



Graphitized Carbon Blacks



Fence line Monitoring EPA325B (Draft)

On September 29, 2015, the U.S. EPA issued a final rule - requiring all U.S. Petroleum Refineries to conduct passive air sampling along the perimeter of their properties. (40 CFR Parts 60 and 63) Coming Method will be EPA325A/B.

EPA Method 325 uses passive (diffusive) samplers to collect air samples at specific intervals along the fence line of the petroleum refineries property¹. The target compound is Benzene.

These passive air samplers are comprised of a inert-coated* stainless steel thermal desorption tube packed with a graphitized carbon adsorbent. Carbopack™X is listed as primary adsorbent.

The Carbopack X is also validated by the EPA and listed in the Method.

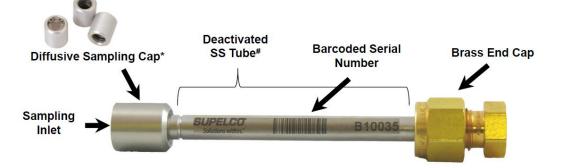
*The I.D. of the stainless steel tube are coated with a protective layer (Supelcoat[™]), which masks any active sites that could be present on the inside of the tube.



EPA Method 325 (DRAFT) for Fenceline Monitoring (FLM)

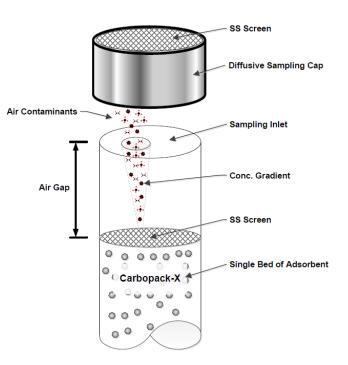
Using **Carbopack X** in Specially Treated Thermal Desorption Tube

- Inert TD Tube with diffusion cap
- Carbopack X bed
 - retains wide range of analytes
 - recovery of key analytes, 1,3-butadiene, benzene, toluene, remain at 100% when sampling large volumes
 - is hydrophobic
 - water will not be retained



Tube Dimensions: 6.35 mm O.D. x 5 mm I.D. x 89 mm Long

Cat# 28686-U Stainless Steel w/SupelCoat Carbopack X (60/80) shown with diffusive sampling cap



Method 325B-Volatile Organic Compounds from Fugitive and Area

Validated Sorbents and Uptake Rates (mL/min) for

Sources:

Sampler Preparation and Analysis

Selected Clean Air Act Compounds

Table 12.1:

Carbopack X was validated for the broadest compound portfolio due to its broad suitability, not just Benzene (or BTEX).

EPA325B (Draft)

Compound	Carbopack™ Xª
1,1-Dichloroethene	0.57±0.14
3-Chloropropene	0.51±0.3
1,1-Dichloroethane	0.57±0.1
1,2-Dichloroethane	0.57±0.08
1,1,1-Trichloroethane	0.51±0.1
Benzene	0.67±0.06
Carbon tetrachloride	0.51±0.06
1,2-Dichloropropane	0.52±0.1
Trichloroethene	0.5±0.05
1,1,2-Trichloroethane	0.49±0.13
Toluene	0.52±0.14
Tetrachloroethene	0.48±0.05
Chlorobenzene	0.51±0.06
Ethylbenzene	0.46±0.07
m,p-Xylene	0.46±0.09
Styrene	0.5±0.14
o-Xylene	0.46±0.12
p-Dichlorobenzene	0.45±0.05

^a Reference 3, McClenny, J. Environ. Monit. 7:248-256. Based on 24-hour duration.

More Information

US EPA Method 325B Compliance Flyer (RNW)

Passive Fenceline Sampling Solution for Benzene and Other VOCs

sigma-aldrich.com/flm

SUPELCO US EPA Method 325B Compliance

Passive Fenceline Sampling Solution for Benzene and Other VOCs



Proven quality and performance you can trust from the inside out with the Fenceline Monitor (FLM[®]) Carbopack[®] X deactivated stainioss steel TD tube

Carbopack X delivers trusted sampling performance from the inside while the easy to read tube markings and barcode delivers confidence and traceability on the outside.

Features and Benefits

- . Only sampling device available with Carbopack X adsorbent
- validated by the US EPA . Durable barcode to ensure traceability and minimize
- litigation risk · Easy-to-read tube markings
- Certificate of Analysis included in every box
- Low-cost monitoring solution, tubes are reusable up to 100x.



Compliance with US EPA Method 325B The new US federal regulation 40 CFR Parts 60 and 63 - Petroleum

Refinery Sector Risk and Technology Review and New Source

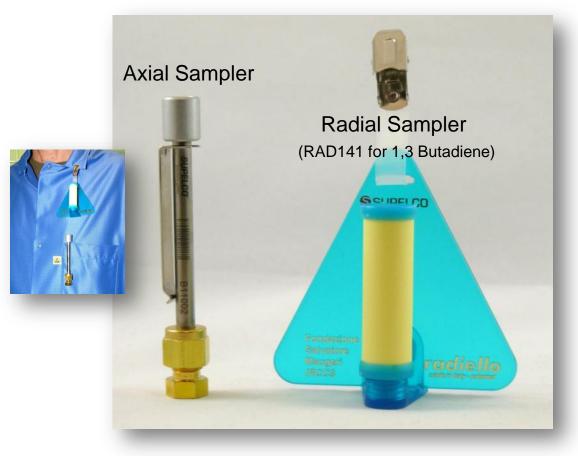
Performance Standards, is an update to the previous version. It requires that the perimeter or fenceline of refineries are monitored. for 14-day periods using passive sampling sorbent tubes at multiple locations for benzene and other volatile organic compounds (VOCs). The Complete Sampling and Analysis Solution Our goal is to bring you the science and performance inside the tube and provide the quality on the outside to help you achieve compliance with the ruling. We make the carbon adsorbents and pack the adsorbent tubes in-house with our adsorbent manufacturing. Tube packing and OC are incated in suites next to each other, no other supplier has the same capability If you perform the laboratory analysis, our GC Capillary Columns are manufactured right next door in another suite. Our RMD facilities and experts are also located onsite

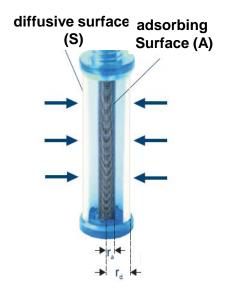
- Only simpling device with simpling rates for Carbopack X





Passive Sampling Options for Thermal Desorption Using Carbopack X



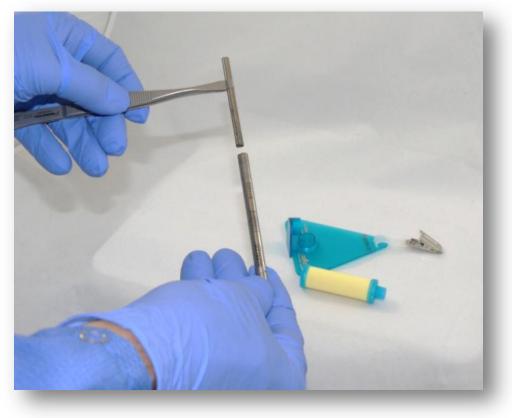


Radial Sampler (radiello®) due to design

- Higher Sampling rates
- Higher sensitivity

Radiello (RAD141) Sampler for 1,3-Butadiene

After sampling, the RAD141 adsorbent cartridge is placed in an empty stainless steel thermal desorption tube for analysis.





Carbon Molecular Sieves

Carbosieve® and Carboxen®

- Maximum temperature: 400 °C
- Recommended desorption temp: 330 °C
- Recommended conditioning temp: 350 °C
- Methanol is retained

Typical Characteristics

- Spherical (Carbosieve-G is granular)
- Shiny/Dull Black in color
- High surface area 400 to 1500 m²/g
- Designed to retain and release small molecular weight compounds



Adsorptive Strength of Carbon Molecular Sieves

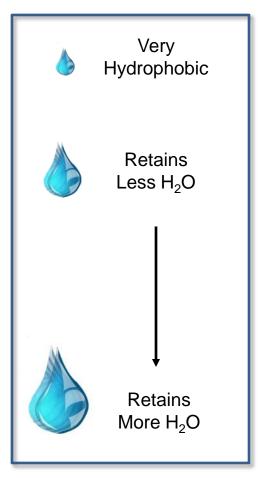
Carboxen-1016 (75 *m*²/g) Carbosieve-G $(1160 \text{ m}^2/\text{g})$ *Carboxen-1012* (1500 m²/g) Carboxen-564 $(400 \text{ m}^2/\text{g})$ Carboxen-1000 (1200 m²/g) Carboxen-1001 (500 m²/g) Carboxen-569 $(485 \text{ m}^2/\text{g})$ Carboxen-1003 (1000 m²/g) Carboxen-1018 (675 m²/g) Carbosieve-SIII (975 m²/g) Carboxen-1021 (1160 m²/g)

Weakest

Strongest

Relative Adsorption Strength

Relative Hydrophobicity



Glass Beads Graphitized Carbon Blacks Porous Polymers

Carboxen-1016 Carboxen-569,1001,1003 Carboxen-563 Carboxen-564 Carboxen-564 Carboxen-1000 Carboxen-1012 Carboxen-1018,1021, Carbosieve G & SIII

Silica Gel, Mole Sieve 5x, 13x

Information on Carbon Adsorbens

Wide Range of Properties How to choose?



Relative Analyte Size ¹	Adsorptive Strength	Recommended Adsorbents (listed highest to lowest surface area)
C20+	Weakest	Carbotrap F, Carbopack F
C12-C20		Carbotrap C, Carbopack C
C9-C14		Carbotrap Y, Carbopack Y
C5-C12		Graphsphere 2027, Graphsphere 2029, Carbotrap B, Carbopack B, Graphsphere 2016, Graphsphere 2017
C3-C9		Carbotrap X, Carbopack X, Carbopack Z
C2-C5	Strongest	Carboxen 1012, Carboxen 1034, Carboxen 1000, Carboxen 1008, Carboxen 1026, Carbosieve G, Carboxen 1005, Carboxen 572, Carbosieve S-II, Carboxen 1003, Carbosieve S-III, Carboxen 1032, Carboxen 1030, Carboxen 1006, Carboxen 1018, Carboxen 1010, Carboxen 1021, Carboxen 563, Carboxen 1001, Carboxen 569, Carboxen 1033, Carboxen 564

	Approx. Surface Area	Ve	Approx. Pore plume (co		Approx Pore Diam.
Carbon	(m2/g)	Micro	Meso	Macro	(Å)
Carbotrap F /	5	-	-	-	-
Carbopack F *					
Carbotrap C /	10	-	-	-	-
Carbopack C *					
Carbotrap Y/	24	-	-	-	_
Carbopack Y *					
Carboxen 1017	61	-	0.33	-	-
Graphitized carbon black	70	0.01	0.23	-	137
Carboxen 1016	75	-	0.34	-	-
Carbotrap B / Carbopack B *	100	-	-	-	-
Mesoporous carbon	203	_	0.49	_	96.3
Purified carbon black	214	0.06	0.28	-	63.9
Carbopack Z *	220	_	1.73	_	255
Carbotrap X /	240	_	0.62	_	100
Carbopack X *					
Carboxen 564	400	0.24	0.13	0.14	6-9
Carboxen 569	485	0.20	0.14	0.10	5-8
Carboxen 1001	500	0.22	0.13	0.11	5-8
Carboxen 563	510	0.24	0.15	0.24	7-10
Carboxen 1021	600	0.30	-	_	5-8
Carboxen 1010	675	0.35	-	-	6-8
Carboxen 1018	675	0.35	-	-	6-8
Carboxen 1006	715	0.29	0.26	0.23	7-10
Carbosieve S-III	975	0.35	0.04	-	4-11
Carboxen 1003	1000	0.38	0.26	0.28	5-8
Carbosieve S-II	1059	0.45	0.01	_	6-15
Carboxen 572	1100	0.41	0.19	0.24	10-12
Supelcarb	1150	0.47	0.26	0.28	5-8
Carbosieve G	1160	0.49	0.02	_	6-15
Carboxen 1000	1200	0.44	0.16	0.25	10-12
Carboxen 1012	1500	_	0.66	_	19-21

Adsorbent / Thermal Desorption Literature

"A Tool for Selecting an Adsorbent for Thermal Desorption"

www.sigmaaldrich.com/air-monitoring

Carbopack C Carbopack Y Carbopack B Carbopack X Carboxen-1000 (Carbon Molecular Sieve) Surface Area: 1200 m²/g n Temperature: 330 °C Challenge Volume (Liters) 10 20 5 lalocarbon 12 hloromethane alocarbon 114 1.3-Butadiene Chloroethane Halocarbon 11 Acrylonitrile Car Carboxen-1000 Carboxen-1001 Methylene chloride 3-Chloropropene Halocarbon 113 roethane cis-1,2-Dichloroethene Chloroform ichloroethane 1,1-Trichloroethane Benzene tetrachloride ichloropropane Frichloroethene cis-1,3-Dichloropropene propropene trans-Toluene 121 Dibromoethane 140 (140 (140 Chlorobenzene Ethylbenzene m & p-Xylene Styrene Carbo Carboxen-1018 Carbosieve S-III o-Xylen 4-Ethyltoluene imethylbenzene nethylbenzen 3-Dichl obenzene -Dichlorobenzene 2-Dichlorobenzen enzene .98 Performance Key Labor. SUPELCO Solutions within.

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Technical Report

A Tool for Selecting an Adsorbent for Thermal Desorption Applications

Research conducted by Jamie Brown, R&D, Co-author Bob Shirey, R&D

There are varieties of adsorbents used in the field of thermal desorption. Often choosing the right adsorbent can be difficult. The goal in selecting the proper adsorbent is to choose one that can retain a specific or group of analytes for a specified sample volume. However, just as important the adsorbent must also be able to release the analyte(s) during the desorption process. This report sheds some light on choosing the right adsorbent by demonstrating the relative differences between those most commonly used. Some of the adsorbents investigated in this research were Tenax TA[®], Carbotraps[™], Carboxens[™], Carbosieve[™], charcoals and glass beads. The test probe for this research was a gas mix containing forty-three different analytes whose physical properties ranged from 50 to 260 in molecular weight and -30 to 215°C in boiling point. The analytes in this mixture are a subset of the EPA Hazardous Pollutant list. EPA method TO-17 is the typical method you use to sample these analytes. We introduced this gas mixture to each of the adsorbents using the flash vaporization technique and then challenged each with various sampling volumes ranging from 0.2 to 100 liters. We thermally desorbed each of the adsorbents into a GC/MSD system.

1	Table	to t	Contents	
1	Abstra	ict .		

ADSEAC
Introduction
Experimental Details
Sequence of Events
Setting Up the Challenge Volume
The Analysis Matrix
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Calculating the Recovery of the Second Desorption
Results: How to Use the Charts
General Guidelines for Interpreting the Trends
Using the Charts to Design a Multi-Bed Tube
Discussion of the Results
Conclusion
Questions and Answers
Acknowledgements
References
Performance Charts



Introduction

Our goal in performing this research was to develop a simple and easy to use too for thermal description users. This 'not' denonstrates the relative difference between the adsorbents based on challenged with various sample volumes. Several other condtradive hundly of the sample are an all influence the abelity of an adsorbent to retain an analyte during the sampling process. This research covers only the sample volume aspect.

The challenge we posed to each of the adsochents was to spike a known quanity of a test mix not be adsochents. Then challenge the adsochent by subjecting it to a constant flow of clean nitrogen until we obtained the desired volume. We then thermally desorbed the adsorbents into a GC system to deternise what analysis eminant of recovered on the adsochent after it we subjected it to the challenge volume. This was repeated for sid different volumes of nitrogen.

An analogy that depicts the challenge posed by the research is that of packed outline rhormologying/t. For this, we pack the adsorbent into a colled column, we apply a cartier gas to carry the detector at the opposite and. Essentially the same concept saids the implementation of the second second second second bent is packed into a empty themat disception tube (very small column). The carter gas for this research was inforced, twey small column). The carter gas for this research was inforced. Twey is the implementation of the second tube integrates and a single second the second second tube integrates the implementation of the second tube integrates at some point in time. some of the analyses beach-through whereas, others are retained by the adsorbent retained. This research locks at what analyses the adsorbent retained. We seek the other is the column of the bub to analyse what broke-through the research locks at what analyses the adsorbent retained. Not swetter for detection.

We are committed to the success of our Customers, Employees and Shareholders through leadership in Life Science, High Technology and Service.

Spiking the Tubes with the Gas Mix

Adsorbent Tube Injector System (ATIS)

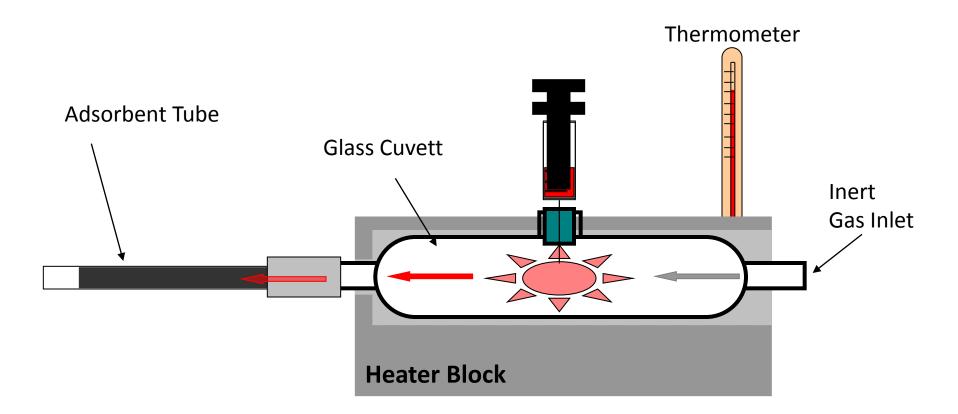
used for spiking tubes

- 20 mL of gas mix
 - 43 Compounds: 50 to 260 in molecular weight, -30 to 215°C in boiling point.
 - Concentration : 1ppm of each compound
- Injected into a stream of N₂
- N₂ carried the compounds to the tube
 - Challenges volumes tested



sigma-aldrich.com/atis

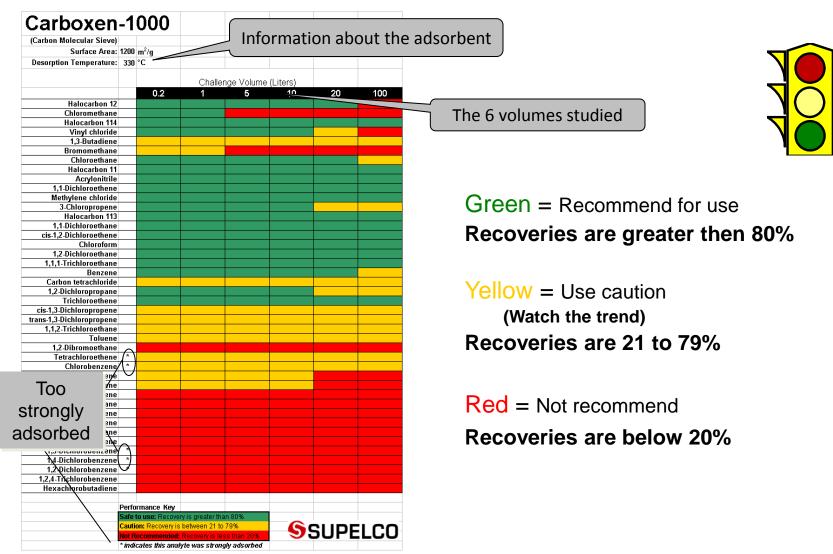
ATIS - Principle



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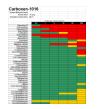
How to use the performance charts



59

The outcome of this research ...

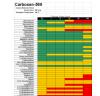


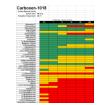


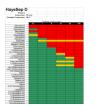




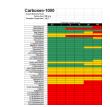
Carbopack C





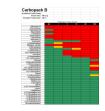


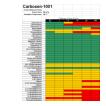


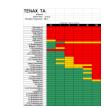








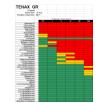








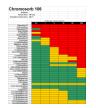




Rathan Sana Mill Dependent Temperature - Mil			
To Management			
1.3 Subsection 7			
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"Inlet" First adsorbent bed

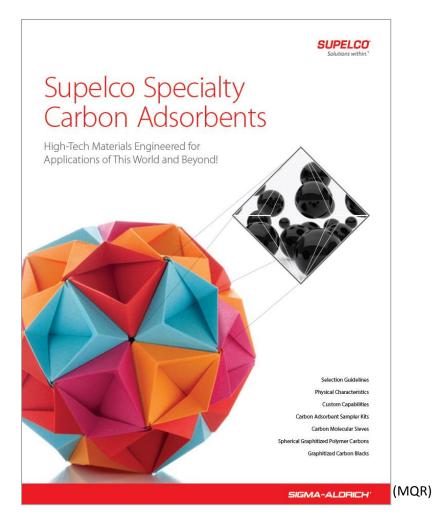
Carbopac	k	В					
(Graphitized Carbon Black)							
Surface Area:	100	m²/g					
Desorption Temperature:	330	°C					
		0.2		nge Volume 5		20	100
Vinyl chloride		0.2				20	100
Methylene chloride			Brea	k s thro u	lgh		
Toluene			Pote	ined			
1,2,4-Trichlorobenzene			Nete	inteu		/	
					\land /		

Back-up adsorbent bed

Carboxen	-1	003					
(Carbon Molecular Sieve)							
Surface Area:	1000	m²/g					
Desorption Temperature:	330	°C					
		0.2		nge Volume 5	e (Liters) 10	20	100
Vinyl chloride			Dete	inad			
Methylene chloride			Reta	inea			
Toluene	(*)	Carryover	was observed (Too Strongly	dsorbed)		
1,2,4-Trichlorobenzene		Most like	ly irreversibly a	dsorbed			

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Test different Carbon Adsobents?



Adsorbent	Type ¹	S/G ²	Sc ³	Approximate						
				Surface Area	Pore Volume (cc/g)			Pore Diameter	Micropore Diameter	
				(m²/g)	Macro	Meso	Micro	(Å)	(Å)	рН
Carboxen 1012	CMS	S		1500	_	0.66	_	_	19-21	_
Carboxen 1034	CMS	S	Y	1260	0.10	0.48	0.42	32	5-20	10.5
Carboxen 1000	CMS	S		1200	0.25	0.16	0.44		10-12	_
Carboxen 1008	CMS	S		1200	0.25	0.16	0.44		10-12	_
Carboxen 1026	CMS	S		1200	_	0.06	0.72		4-20	_
Carbosieve G	CMS	G		1160	_	0.02	0.49		6-15	_
Carboxen 1005	CMS	S	Y	1150	0.28	0.26	0.47	_	5-8	9.3
Carboxen 572	CMS	S	Y	1100	0.24	0.19	0.41	_	10-12	9.5
Carbosieve S-II	CMS	S		1059	_	0.01	0.45	_	6-15	_
Carboxen 1003	CMS	S	Y	1000	0.28	0.26	0.38	_	5-8	9.2
Carbosieve S-III	CMS	S		975	_	0.04	0.35	_	4-11	_
Carboxen 1032	CMS	S	Y	820	0.10	0.38	0.29	37	4-20	3.0
Carboxen 1030	CMS	S	Y	740	0.11	0.13	0.26	26	5-20	4.0
Carboxen 1006	CMS	S		715	0.23	0.26	0.29	_	7-10	_
Carboxen 1018	CMS	S		675	_	_	0.35		6-8	_
Carboxen 1010	CMS	S		675	_	_	0.35		6-8	_
Carboxen 1021	CMS	S		600	_	_	0.30		5-8	_
Carboxen 563	CMS	S	Y	510	0.24	0.15	0.24	_	7–10	6.8
Carboxen 1001	CMS	S		500	0.11	0.13	0.22	_	5-8	_
Carboxen 569	CMS	S	Y	485	0.10	0.14	0.20	_	5-8	8.6
Carboxen 1033	CMS	S	Y	420	0.10	0.10	0.15	33	5-17	7.0
Carboxen 564	CMS	S	Y	400	0.14	0.13	0.24		6–9	8.7
Carbotrap X	GCB	G	Y	240	_	0.62	_	100	_	9.5
Carbopack X	GCB	G		240	_	0.62	_	100		_
Carbopack Z	GCB	G		220	_	1.73		255		_
Purified Carbon Black	CB	G		214	_	0.28	0.06	63.9	_	_
Mesoporous Carbon	GCB	G		203	_	0.49		96.3		_
Graphsphere 2027	SGPC	S		126	0.20	0.35	_	173	_	_
Graphsphere 2029	SGPC	S		105	0.20	0.26	0.03	180	_	_
Carbotrap B	GCB	G	Y	100	_	_	_	_	_	9.7
Carbopack B	GCB	G		100	-	_	_	_	_	_
Graphsphere 2016	SGPC	S		75	_	0.34	_	_	_	_
Graphitized Carbon Black	GCB	G		70	_	0.23	0.01	137	_	_
Graphsphere 2017	SGPC	S	Y	60	_	0.33				
Carbotrap Y	GCB	G	Y	24	_	_		-	mes	1
Carbopack Y	GCB	G		24	_	_	000	O		11-

Custom Tubes possible!

G

G

G

G

Y

10

10

5

GCB

GCB

GCB

GCB

Carbotrap C

Carbopack C

Carbotrap F

Carbopack F

Sampler Kits

Conclusion

- Thermal desorption covers a wide range of analytes
 - Available Adsorbents offer enables broad sampling scope
- Single bed tubes are often have limits for a wider analyte portfolio
 - Multibed tubes provide a wider range
- Synthetic carbon adorbents (Carbopack, Carboxen) are most suitable due to stability, reproducibility & purity
 - Sigma-Aldrich / Supelco has long year experience and wide selection
- Adsorbent selection can be done
 - by literature research (official methods, journal articles, vendor information)

- using "A Tool for Selecting an Adsorbent for Thermal Desorption"
- Custom selected adsorbents for a multibed tube allow to achieve optimal retention & release characteristics for desired application

Acknowledgments

Jamie Brown – Principal R&D Scientist William Betz - Associate Fellow R&D Scientist Olga Shimelis - Principal R&D Scientist Kristen Schultz - Air Monitoring Product Manager



Bellefonte, Pennsylvania USA

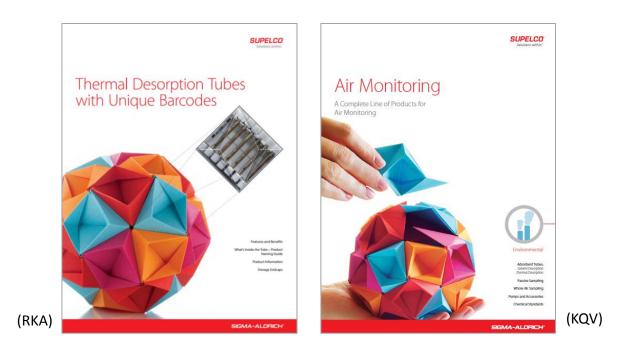
Trademarks: Carbotrap®, Carbopack™, Carboxen®, Carbosieve® – Sigma Aldrich, USA Chromosorb® - Imerys Minerals California, Inc. USA HayeSep® - Hayes Separations, USA PoraPak™ - Waters Associates, USA Tenax® - Buchem B.V., Netherlands



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Thank you!



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