High-precision oxygen concentration analysis using a novel high-temperature pyrolysis technique

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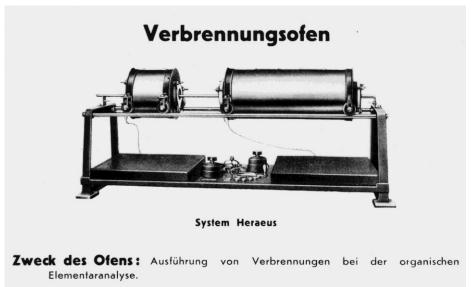


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110 years tradition... Heraeus







The elementar group — tradition means trust



Approx. 1860: W. C. Heraeus utilizes high temperature technologies in order to

process platinum

Approx. 1900: Heraeus produces high purity quartz glass which is soon used in

elemental analysis



Wilhelm Carl Heraeus



Pharmacy "White Unicorn"

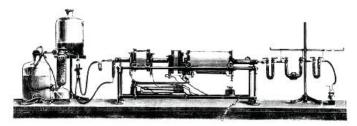
The elementargroup — **tradition means trust**





• Approx. 1905:

First generation of elemental analyzers with electrical furnaces produced by Heraeus

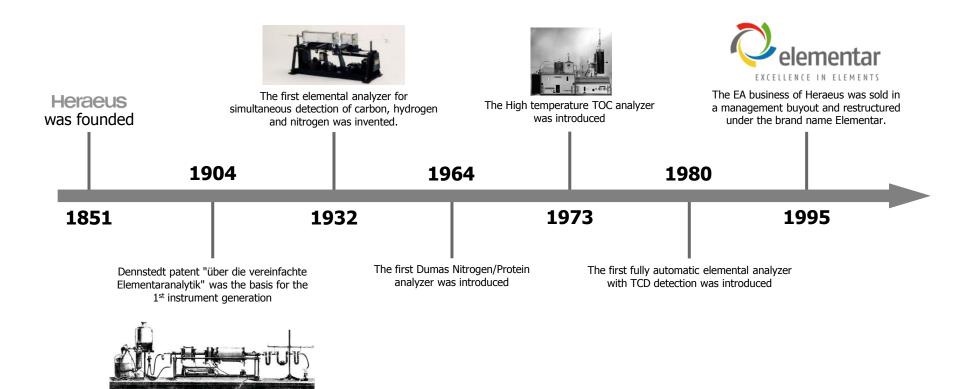


1923:

Fritz Pregl receives the Nobel Prize in Chemistry for his invention of the *method* of micro-analysis of organic substances, using dedicated analytical equipment manufactured by Heraeus!

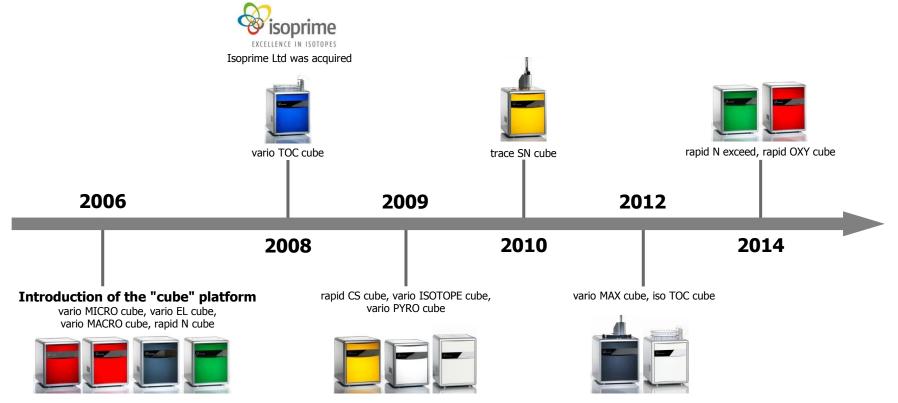
The elementar group history





The elementar group history





The elementar group history





rapid MAX N exceed, rapid MiCRO N cube

2015

Introduction of the inductar series

inductar CS cube, inductar ONH cube, inductar EL cube



The elementar group — about us



elementargroup





- The technology leader in
 - Elemental Analysis (EA)
 - Isotope Ratio Mass Spectrometry (IRMS)

- Offers an unmatched combination of
 - innovative technologies,
 - tailor-made solutions and
 - comprehensive support.

The elementar group – key markets















Chemicals

Agriculture

Energy

Forensics

Materials Environmental



Some of our customers



































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Why Oxygen Analysis?





- Oxygen is an important parameter to describe:
 - Coke and coal (ASTM D3176)
 - Gasoline and methanol fuels (ASTM D5622)
 - Polymers (fire/decomposition resistance)
 - Fine chemicals (quality control)

- Two approaches:
 - Indirect: 100% CHNS% ash% = 0%
 - Direct measurement

Indirect Oxygen Analysis



• Indirect approach: 100% - CHN% - S% - ash% = O%

- Drawbacks:
- Time and resources consuming
- Summation of errors
- Up to three instruments required

"The result so obtained is affected by errors incurred in the other determinations of the ultimate analysis and also by the changes in weight of the ash-forming constituents of ignition." (ASTM D3176)

Direct Oxygen Analysis



New applications require more precise, direct determination of oxygen

"The presence of oxygen-containing compounds in gasoline can promote more complete combustion, which reduces carbon monoxide emissions. The Clean Air Act (1992) requires that gasoline sold within certain, specified geographical areas contain a minimum percent of oxygen by mass (presently 2.7 mass %) during certain portions of the year. The requirement can be met by blending compounds such as methyl tertiary butyl ether, ethyl tertiary butyl ether, and ethanol into the gasoline." (ASTM D5622-95)

> Increasing need for a direct and precise determination of oxygen concentrations, especially in the solid and liquid fuel industry

Direct Oxygen Analysis: Reductive Pyrolysis



- Sample is pyrolyzed in helium atmosphere in presence of carbon black
 - Sample oxygen (O) is converted to carbon monoxide (CO)
 - CO is detected
 - Boudouard equilibrium:

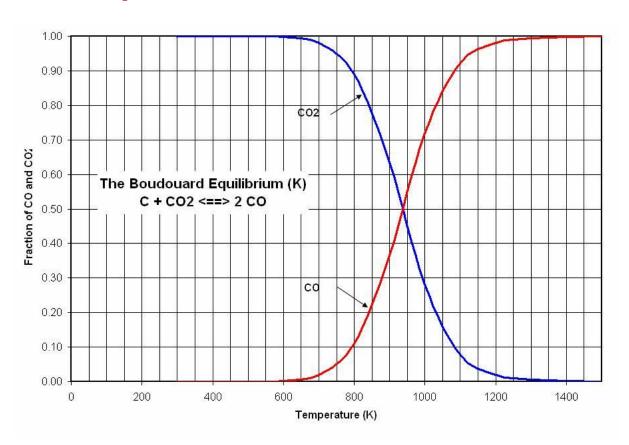
$$CO_2 + C \rightarrow 2 CO$$

2 $CO \rightarrow CO_2 + C$

Only quantitative conversion to CO results in matrix-independence

Boudouard euqilibrium





We want CO! Simplified Pyrolysis Chemistry



The higher the temperature, the more CO!

• Temperatures around 1200° C for sub-percent CO₂ formation

• But: Temperatures > 1200° C require unique technical solutions!

Depending on temperature, a mixture of CO and CO₂ is produced

→ biased oxygen measurement (matrix effects!)

Get OXYted about oxygen analysis - rapid OXY cube -





- Temperature 1450° C
- Matrix independent, no CO₂ formation
- No matrix standards needed
- Excellent, highly accurate results
- Ready to go, stand-alone analyzer

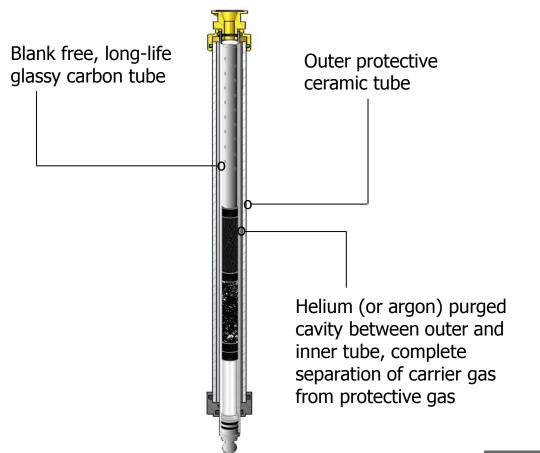


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Get OXYted! - Blank free pyrolysis reactor -

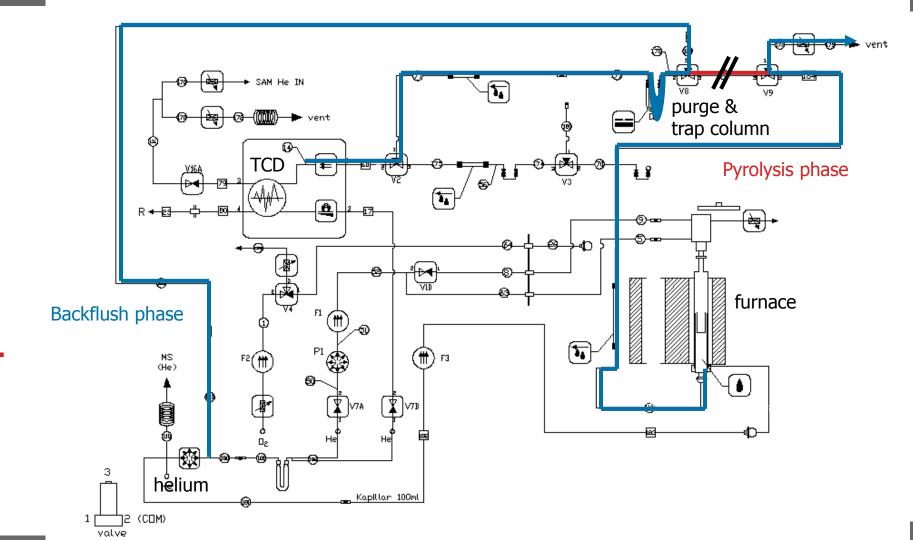




Get OXYted! - Patented backflush -

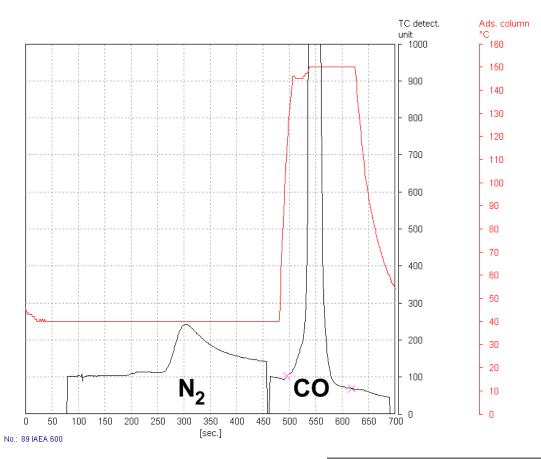


- **Problem:** Pyrolysis is not as well defined as combustion. Interferences can lead to wrong results.
- Function of backflush system:
 - Adsorption of CO during pyrolysis phase
 - Desorption with clean helium and temperature ramp
 - Tailing from pyrolysis byproducts does not enter the TCD but goes to vent
 - Focused peak due to temperature ramp
- Result: Full baseline separation of CO from interfering gases and full recovery



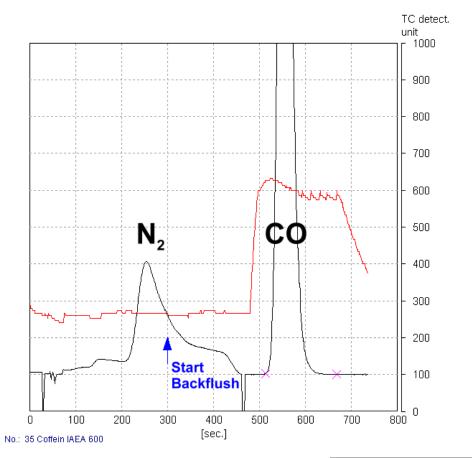
Analysis of caffeine without backflush

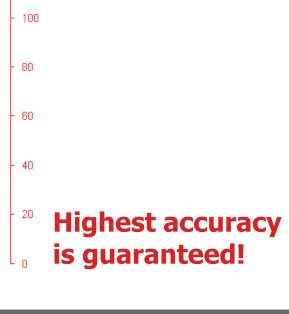




Analysis of caffeine with backflush







CO column

140

120



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Oxygen analysis has never been more reliable



SAMPLE	0 [%]	ABS. SD [%]
OIL 1	0,34	0,03
OIL 2	0,66	0,04
COKE 1	3,93	0,05
COKE 2	1,88	0,10
COAL	6,32	0,08
CAFFEINE	16,94	0,07
ACETANILIDE	11,97	0,05
SULFANILIC ACID	27,77	0,07
SIVER PHOSPHATE	15,35	0,15
BARIUM SULFATE	27,72	0,44
FLUOROBENZOIC ACID	23,50	0,12

rapid OXY cube liquid injection

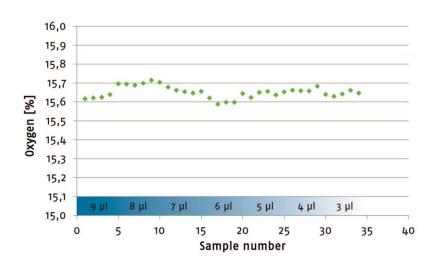


• rapid OXY cube + VLS



Liquid samples: Proof of principle





 34 consecutive hexanol injections over wide range of injection volumes (3-9 μl)

• Theoretical value: 15.66%

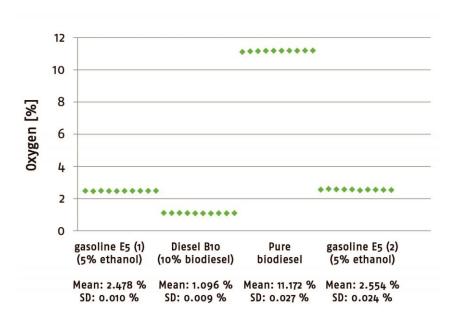
• Mean: 15.65%

• Abs. SD: 0.03%

- Excellent results
- Excellent repeatability
- Excellent accuracy

Oxygen Content in Gasoline Blending





- Oxygen content corresponds to unwanted CO emission in engines
- Minimum oxygen content is legal requirement (e.g. Clean Air Act)
- Pure fossil gasoline with low oxygen content is blended with high oxygen content biofuels
- Excellent repeatability
- No memory effects



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rapid OXY 🍘 cube Get OXYted about oxygen analysis rapid OXY 🍘

Highlights



Highest accuracy and data reliability

 Patented backflush for blank-free oxygen analyses

Designed for unattended 24/7 operation

 Unmatched operating comfort through sophisticated self-diagnosis

High robustness – low maintenance





Elementar thanks for your attention!

that we are a strong family.













